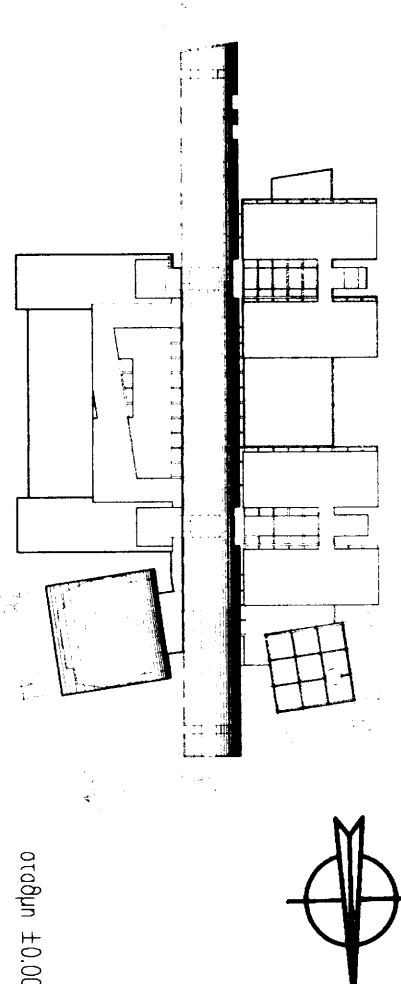




ANAB.	HAMEPOMAHVIA	ΣΥΣΛΙΑΣΗ	ΕΛΕΓΧΟΣ	ΕΓΚΡΙΣΗ	ΠΕΡΙΠΛΗΘΗ
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KAEWA



ΚΤΙΡΙΑΚΕΣ ΕΓΚΑΤΑΣΤΑΣΕΙΣ ΤΜΗΜΑΤΩΝ
ΕΘΝΟΛΟΓΙΑΣ – ΙΣΤΟΡΙΑΣ ΚΑΙ ΦΙΛΟΛΟΓΙΑΣ

[illegible]

ΣΤΑΘΟΣ
ΜΕΤΕΩΡΟΛΟΓΗΣ Η/Μ ΕΚΚΑΤΑΣΤΑΣΕΩΝ

ΜΕΝΕΤΗ ΕΦΑΡΜΟΓΗΣ Η/Μ ΕΚΚΑΤΑΣΤΑΣΕΩΝ

ΘΕΜΕΛΙΑΚΗ ΓΕΙΩΣΗ ΚΤΙΡΙΟΥ 1

address: 40		ΔHE/754		april: A0		KINMAA		1-100	
WEIGHT: 1M		0.034.00		average		APPROX. 240000		address: 1010	
TICK MARK: 1M		07/04.00		average		1010		1010	
ENERGY: 1M		10.04.00		average		1010		1010	

<p>6. <i>Explain the difference between a strong and a weak acid.</i></p> <p>7. <i>Explain the difference between a strong and a weak base.</i></p> <p>8. <i>Explain the difference between a strong and a weak electrolyte.</i></p> <p>9. <i>Explain the difference between a strong and a weak acid.</i></p> <p>10. <i>Explain the difference between a strong and a weak base.</i></p> <p>11. <i>Explain the difference between a strong and a weak electrolyte.</i></p> <p>12. <i>Explain the difference between a strong and a weak acid.</i></p> <p>13. <i>Explain the difference between a strong and a weak base.</i></p> <p>14. <i>Explain the difference between a strong and a weak electrolyte.</i></p> <p>15. <i>Explain the difference between a strong and a weak acid.</i></p> <p>16. <i>Explain the difference between a strong and a weak base.</i></p> <p>17. <i>Explain the difference between a strong and a weak electrolyte.</i></p> <p>18. <i>Explain the difference between a strong and a weak acid.</i></p> <p>19. <i>Explain the difference between a strong and a weak base.</i></p> <p>20. <i>Explain the difference between a strong and a weak electrolyte.</i></p>	<p>Teacher</p> <p>Signature</p>
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